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Mechanical properties and interfacial characteristics of 6061 Al alloy plates fabricated by hot-roll bonding

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Abstract: This work aims to investigate the mechanical properties and interfacial characteristics of 6061 Al alloy plates fabricated by hotroll bonding (HRB) based on friction stir welding. The results showed that ultimate tensile strength and total elongation of the hot-rolled and aged joints increased with the packaging vacuum, and the tensile specimens fractured at the matrix after exceeding 1 Pa. Non-equilibrium grain boundaries were formed at the hot-rolled interface, and a large amount of Mg₂Si particles were linearly precipitated along the interfacial grain boundaries (IGBs). During subsequent heat treatment, Mg₂Si particles dissolved back into the matrix, and Al₂O₃ film remaining at the interface eventually evolved into MgO. In addition, the local IGBs underwent staged elimination during HRB, which facilitated the interface healing due to the fusion of grains at the interface. This process was achieved by the dissociation, emission, and annihilation of dislocations on the IGBs.

Keywords: 6061 Al alloy; hot-roll bonding; vacuum; mechanical properties; interfacial grain boundaries

1. Introduction

Due to good specific strength, corrosion resistance, weldability, and formability, light alloys represented by 6xxx Al alloy are widely used in aerospace, high-speed ships, construction profiles and other fields [1–2]. The corresponding key components often need to be prepared using large Al alloy ingots with fine grains and uniform composition. However, defects such as severe macro-segregation, porosity, and cracks often occur in extra-large ingots produced using traditional methods of direct chill (DC) casting, which seriously reduces the yield and comprehensive performance of Al alloys [3–5]. And these ingots require prolonged homogenization annealing to eliminate dendrite segregation and extensive milling after subsequent hot rolling due to the crocodile mouth [6], which results in higher energy consumption and lower yield.

Currently, solid phase bonding technology has been widely used in the industrial field [7–8], and high-quality heavy forgings and ultra-heavy steel plates have been successfully produced by hot-compression bonding (HCB) [9–10] and vacuum cladding rolling (VCR) [11], i.e., the preparation of heavy plates or forgings using solid phase bonding of thin plates for thickening. For Al alloys, Hazelett continuous casting direct rolling (CCDR) process has been able to produce continuous casting billets with a thickness of about 19 mm after years of research and development [12–13]. Compared to traditional DC casting, defects such as

macroscopic segregation, porosity and cracks can be effectively avoided because the cooling rate from the surface to the core is more uniform. Moreover, the thin plates produced by CCDR do not require a homogenization process and extensive milling, which greatly reduces the production cost and energy consumption of Al alloy thin plates [12,14]. Based on the above technological achievements, this study introduced hot-roll bonding (HRB) into the field of Al alloy preparation, i.e., thick Al plates were prepared by roll-bonding on continuous casting billets. It is worth noting that the Al surface was extremely susceptible to the formation of oxide film that were highly adhesive, chemically stable and insoluble in the aluminum substrate, which thickened rapidly with increasing temperature [15]. Tang et al. [16] conducted friction-based additive manufacturing on Al-Mg-Si alloy under atmospheric pressure and found obvious weak-bonding defects at the interface. Therefore, vacuum conditions played an important role in inhibiting the thickening of surface oxide films. For this issue, vacuum friction stir welding (FSW) equipment has been developed for vacuum packaging due to the poor performance of electron beam welding of Al alloys [17]. In fact, excessive vacuum conditions can lead to higher process costs, so it is necessary to carry out HRB experiments at different vacuum conditions to determine a relatively suitable process window.

For the solid phase bonding of Al alloys, many scholars have conducted related studies. Xu *et al.* [18] performed hot pressing bonding on 2A12 Al alloy and found that the inter-

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face healing was achieved by dynamic recrystallization (DRX) induced by the migration of interfacial grain boundaries (IGBs). Zhang et al. [19] stated that discontinuous dynamic recrystallization (DDRX) and geometric dynamic recrystallization (GDRX) had a positive effect on interface healing during vacuum roll cladding of 7050 Al alloy. Xu et al. [20] also reported that DDRX induced by grain boundaries (GBs) bulging or particle-stimulated nucleation contributed to interface healing of 2196 Al alloy during HCB. Yu et al. [21] found that the interface bonding effect of 6063 Al alloy was positively correlated with hot-pressing temperature, and the optimal bonding effect was obtained at 530°C. Furthermore, Xie *et al.* [10] found that the bonding interface of 316LN stainless steel was gradually eliminated during HCB, followed by the fusion of grains on both sides of the interface, which meant that the healing of the interface originated from the elimination of IGBs. Therefore, it is necessary to investigate the interface healing mechanism during HRB of 6061 Al alloy.

In addition, 6061 aluminum alloy is a heat treatable strengthening alloy, and the corresponding aging precipitation sequence is as follows: SSSS (super saturated solid solution) \rightarrow Atomic clusters (Mg, Si) \rightarrow GP zones (nano-precipitation phase) $\rightarrow \beta''$ (Mg₅Si₆) [22] $\rightarrow \beta'$ (Mg₅Si₅) [23] $\rightarrow \beta$ (Mg₂Si, equilibrium phase) [24]. During hot rolling, the second-phase particles in the matrix are heavily precipitated and coarsened, losing the strengthening effect on the matrix. Solution and aging treatments are required to regulate the

size, morphology and quantity of strengthening phase particles in the matrix to further regulate the comprehensive performance of the Al alloy plate after hot rolling [25]. Therefore, it is also necessary to study the microstructure and mechanical properties of the aged interface of 6061 Al alloy.

In this work, 6061 Al alloy was subjected to HRB, solution and aging treatment under different vacuum, and the microstructure evolution and mechanical properties at the interface of Al alloy plates were investigated.

2. Experimental

6061-T6 Al alloy plates with the dimensions of 250 mm × 250 mm × 25 mm were used in this work, and chemical compositions of the materials are given in Table 1. Fig. 1(a) shows the schematic of HRB. Surface treatment must be carried out before symmetrical assembly to expose the fresh metal. The 3D surface topography of the treated plates is shown in Fig. 1(b), and the corresponding roughness value (R_a) was 0.53 µm. The treated plates were assembled on the mechanical device inside the vacuum cabin [19] and welded under different vacuum (10^5 , 10^3 , 1, and 10^{-2} Pa), as shown in Fig. 1(c). Subsequently, HRB with a rolling reduction of 80% was carried at 530°C. The solution and aging treatment process was 540°C × 1 h + 177°C × 8 h. Schematic of sampling location and tensile sample shape on hot-rolled and aged plates is shown in Fig. 1(d).

Table 1. Composition of materials used								wt%
Mg	Cu	Si	Fe	Ni	Cr	Mn	Zn	Ti
1.02	0.245	0.688	0.459	0.31	0.23	0.12	0.08	0.03



Fig. 1. (a) Schematic of HRB, (b) 3D surface topography after surface treatment, (c) assembly and welding in vacuum cabin, and (d) schematic of sampling location and tensile sample shape.

The samples were cut at the interface of the hot-rolled and aged plates using wire electrical discharge machining, followed by standard metallographic procedures. The bonding zone was observed using OLYMPUSDSX500 optical microscope (OM) and electron probe micro-analyzer (EPMA, JXA-8530F, JEOL) after polishing. Phase identification of the hot-rolled and aged samples was done by X-ray diffraction (XRD, PANalytical, XPertpro) with Cu K_{α} radiation. Furthermore, the high-temperature confocal laser optical microscope (CLOM, VL2000DX-SVF17SP/15FTC) was used to in situ observe the precipitation and dissolution process of Mg₂Si during solution process, and corresponding heating process was from room temperature to 500°C at 30°C·s⁻¹, then to 540°C at 10°C·min⁻¹, and hold for 5 min. To further investigate interfacial characteristics, EBSD analysis was carried out using a Zeiss ultra-55 scanning electron microscope (SEM) with the step of 0.1 μ m.

3. Results and discussion

3.1. Evolution of interfacial microstructures

Fig. 2 shows cross-sectional OM images of the hot-rolled and aged interface at different vacuum. It can be seen that there are discontinuous gaps and obvious inclusion defects at atmospheric pressure as the joints do not achieve complete physical contact, which leads to the unhealed interface, as shown in Fig. 2(a) and (b). When the vacuum increased to 10^3 Pa, local healing occurred at the interface, although the bonding interface still showed obvious traces (Fig. 2(c) and (d)). As the vacuum increased to 1 Pa and 10^{-2} Pa, there were fine black precipitation phase particles intermittently distributed along the hot-rolled interface (Fig. 2(e) and (g)). After solution and aging treatment, these precipitated phase particles dissolved back into the matrix and eventually some interfacial products were remained at the interface (Fig. 2(f) and (h)).



Fig. 2. OM images of the hot-rolled (a, c, e, g) and aged (b, d, f, h) interface at 10^5 Pa (a, b), 10^3 Pa (c, d), 1 Pa (e, f), and 10^{-2} Pa (g, h).

In order to investigate the microstructure evolution of 6061 Al allov joints, the element distribution and phase composition of the hot-rolled and aged bonding interface at 10^{-2} Pa were detected, and the corresponding results are shown in Fig. 3. Table 2 shows that the energy spectrum results of the corresponding points in the hot-rolled microstructure. Many elliptical black granular precipitates distributed intermittently along the hot-rolled bonding interface (about 1 µm along RN and 0.6 µm along ND), as shown in Fig. 3(a), and the energy spectrum results indicated that such precipitates were rich in Mg, O, and Si elements. In addition, it can be seen that there were also black precipitates with similar morphology, size, and element distribution in the matrix, indicating that the precipitates at the interface were the same as the black precipitated particles in the matrix during HRB. The aged interface showed a linear healing interface locally, and the precipitated particles at the interface have completely disappeared (Fig. 3(b)), suggesting that the precipitated phase at the hot-rolled interface dissolved during the subsequent heat treatment process.

XRD pattern at the hot-rolled and aged interface at 10^{-2} Pa was shown in Fig. 4, and the corresponding results indicated that there were mainly three phases in the hot-rolled microstructure (Fig. 4(a)), namely α -Al, Mg₂Si (β), and Mg₅Si₆ (β''). Qian *et al.* [22] analyzed the formation mechanism of β'' phase through transmission electron microscope (TEM) and first principles calculations, and found that β'' phase was only maintained at the nanoscale. Therefore, the larger black particle phase precipitated at the interface was Mg₂Si (Fig. 3(a)). During the sample preparation process, Mg₂Si phase absorbed O from the air due to its high Mg-rich content, resulting in a higher content of O in EDS detection results, and Peng et al. [2] also found a similar phenomenon. The white particle phase in the hot-rolled matrix was an insoluble impurity phase contained Fe element (Fig. 3(a)), i.e., α -Al(Mn, Fe)Si, which was not detected by XRD due to its relatively low content. The element distribution at the aged interface indicated that the interface contained trace amounts of Mg and O elements (Fig. 3(b)), indicating that Al₂O₃ at the interface eventually evolved into Mg-containing oxides.

The ΔG -T relationship in Fig. 5(a) indicates the oxidation



Fig. 3. Element distribution at the hot-rolled (a) and aged (b) interface at 10^{-2} Pa.

Tuble 2. Energy dispersive speed occopy (EDS) analysis (were) on the selected positions in Fig. o							
Spectrum	Al	Mg	Si	0	Mn	Fe	Identification
1	84.77	4.03	4.92	5.90	0.38	0	Mg ₂ Si
2	85.31	0	3.61	0	2.75	8.33	α-Al(Mn, Fe)Si
3	98.47	0.93	0.60	0	0	0	Matrix
4	84.42	3.74	5.39	6.10	0.35	0	Mg ₂ Si
20000 16000 12000 12000 8000 4000	$ \begin{array}{c} 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 $	$(100) \rightarrow (100) \rightarrow (100$	$\begin{array}{c} \text{Si}_{5}\text{Si}_{6} \\ \text{(1220)} \\ \text{(1220)} \\ \text{(111)} \\ \text{(2210)} \\ ($	20000 16000 in tr 12000 8000 4000 0	(b) (b) (c) (c) (c) (c) (c) (c) (c) (c) (c) (c	40 50 (111) ◆ (111) ◆ (200) ◆ (200) ◆	$\begin{array}{c} \bulletAl \\ \bulletMg_2Si \\ \hline 022Si \\ \bullet \\ \bullet \\ \hline 00000000 \\ \bullet \\ \bullet \\ \hline 000000000 \\ \bullet \\$
	20 30 40	50 6 $2\theta / (^{\circ})$	0 70 8	0	20 30	40 50 $2\theta/(^{\circ})$	60 70 80

Table 2. Energy dispersive spectroscopy (EDS) analysis (wt%) on the selected positions in Fig. 3

Fig. 4. XRD pattern at the hot-rolled (a) and aged (b) interface at 10^{-2} Pa.

reactions of the corresponding metals can all proceed spontaneously due to $\Delta G < 0$ [26–27]. During solid-state bonding, the oxide layer at the interface is amorphous Al₂O₃ with a thickness of about 5–10 nm. According to the theory of selective oxidation, the greater the affinity with O, the more easily oxides can be formed [26]. Therefore, amorphous Al₂O₃ film can react with Mg to form crystalline MgO as per the chemical reaction (1) in the presence of Mg in the alloy [28]. Panda *et al.* [29–30] investigated the chemical composition of the oxide film on the surface of Al–Mg alloys using real-time *in situ* spectroscopic ellipsometry (RISE) and angle-resolved X-ray photoelectron spectroscopy (AR-XPS), and quantitatively confirmed that Al₂O₃ would gradually evolve into MgO in the range of 125–200°C. In addition, MgO further evolved into MgAl₂O₄ spinel with Al₂O₃ as per the chemical reaction (2) [31]. ΔG –T relationship about Al₂O₃ in the range of 100–700°C is shown in Fig. 5(b) [27]. However, Si *et al.* [31] revealed that the complete evolution of Al₂O₃ to MgAl₂O₄ spinel requires a temperature of 1550°C. However, the high-temperature treatment process involved in this experiment did not exceed 540°C. Therefore, the final evolution product (Mg-containing oxides) at the interface should be MgO (Fig. 3(b)). And this result and the corresponding evolution process have been confirmed by many researchers [27–28,32]. However, due to the extremely low content of MgO, only two phases were detected in the aged microstructure by XRD (Fig. 4(b)), namely α -Al, Mg₂Si (β).



Fig. 5. The ΔG -T relationship curves of (a) oxidation reactions and (b) Al₂O₃ evolution reactions [27]. Reprinted from *Mater. Sci. Eng. A*, 860, X Zhang, Z. A. Luo, Z.S. Liu, *et al*, Interfacial oxide evolution and mechanical properties of 7050 aluminum alloy clad plates during solution and aging process, 144310, Copyright 2022, with permission from Elsevier.

$$3Mg + Al_2O_3 \rightarrow 3MgO + 2Al \tag{1}$$

$$MgO + Al_2O_3 \rightarrow MgAl_2O_4 \tag{2}$$

To investigate the evolution process of precipitate (β) at the interface (marked by microhardness points), high temperature in-situ observation was conducted on the aged joint prepared at 10^{-2} Pa, as shown in Fig. 6. The results show that it is relatively difficult to identify the bonding interface around 200°C (Fig. 6(a)), and intermittently distributed light gray precipitates gradually appeared at the interface after reaching 470°C (Fig. 6(b)). Subsequently, the precipitated phase gradually became clear and distributed along the interface in the form of black particles (Fig. 6(c)). Dolan et al. [33] reported that the quenching sensitivity region range of 6061 Al alloy was 250–450°C, which meant that β phase would precipitate and grow rapidly within this temperature range. However, the coarsening temperature of β phase at the interface was relatively high in this experiment, which may be due to the lag of the actual temperature of the sample caused by the rapid heating rate. This indicated that the bonding interface of the 6061 Al alloy formed the so-called "non-equilibrium" GBs [34], namely IGBs, during HRB. Such nonequilibrium GBs served as rapid diffusion paths of solute atoms [35] and provided nucleation sites for β phase. During



Fig. 6. *In-situ* precipitation and dissolution process of Mg₂Si at the interface: (a, b) below 500°C and (c, d) above 500°C.

the hot rolling and cooling processes, the solute atoms on both sides of the interface underwent strong segregation at IGBs [36], which promoted the formation and growth of β phase. During the solution process, the solubility of the alloy increased with the increasing temperature, and β phase at the interface gradually dissolved back into the matrix (Fig. 6(d)).

3.2. EBSD analysis of the IGBs

To further investigate the bonding behavior of 6061 Al alloy during HRB, the grain structure of the hot-rolled interface (marked by microhardness points) was analyzed, and the corresponding EBSD results are shown in Fig. 7. From inverse pole figure (IPF) map (Fig. 7(a)), it can be seen that the hot-rolled grains were significantly elongated, mainly in [101] orientation, and complete healing occurred at the local bonding interface, i.e., IGBs disappeared and the grains merged, as shown in region 4. Relevant literature indicated that DDRX and GDRX mechanisms contributed significantly to the solid-phase bonding of Al alloys, mainly manifested in the migration and elimination of IGBs [19-20,37]. Therefore, it is necessary to study the DRX behavior at the hot-rolled interface. Grain orientation spread (GOS) can be considered as a measure for dislocation density and strain in individual grains, and grains with GOS value below 1.8 can be defined as recrystallization [38]. From the GOS map (Fig. 7(b), it can be seen that the grains corresponding to region 4 in Fig. 7(a) was not a recrystallized grain. In addition, although there were DRX grains on both sides of the interface, they did not seem to play a critical role in interface healing, suggesting that another healing mechanism might exist in the HRB process of 6061 Al alloy.

In this study, grain boundaries within 2° to 10° , 10° to 15° , and 15° to 60° were defined as low angle grain boundaries



Fig. 7. IPF maps (a), GOS maps (b), and KAM maps (c) at the hot-rolled interface at 10⁻² Pa, the enlarged image (d) of some bonding areas in (a).

(LAGBs, gray), middle angle grain boundaries (MAGBs, red), and high angle grain boundaries (HAGBs, black), respectively. Kernel average misorientation (KAM) map (Fig. 7(c)) reflected that the strain at the interface was basically the same as that of the matrix. Relevant literature indicated that the main DRX mechanism in 6xxx aluminum alloy was continuous dynamic recrystallization (CDRX) [39], which was mainly manifested as follows: First, a large number of dislocations generated by deformation entangled and formed dislocation arrays and dislocation cells (Fig. 7(c)), which developed into LAGBs in the process of continuous dislocation absorption, and then, LAGBs continued to absorb dislocations to further grew into MAGBs, and finally, subgrains gradually rotated into HAGBs. However, a phenomenon similar to the evolution of reverse CDRX was discovered for the first time at the bonding interface, as shown in Fig. 7(a)and (d), i.e., the bonding interface transformed from HAGBs to MAGBs to LAGBs and finally disappeared completely. The analysis results of the corresponding stages indicated that the misorientation along L1–L4 were about 58.6, 13.9, 8.6, and 0.12, respectively, as shown in Fig. 7(d), indicating that the compatibility between the grains on both sides of the interface gradually increased during HRB. Furthermore, an important evidence was found from Fig. 7(a) and (d) that there were IGBs connected by LAGBs, MAGBs and HAGBs at the interface. This also indicated that the bonding interface underwent staged elimination and eventually formed complete grains, as shown in Fig. 7(a) and (d).

As mentioned earlier, the evolution of grain boundaries during thermal deformation is closely related to dislocations. Dislocations in crystals could be divided into two categories: statistically stored dislocation (SSD) and geometrically necessary dislocation (GND) [40-41]. SSD evolves from random trapping processes, and it is currently difficult to directly obtain its density through relevant tests. GND can adjust the curvature of the crystal lattice and reflect strain gradient field. The relevant literature showed that GND could give a representative trend of the total dislocation density in FCC [41–42]. GND density (ρ_{GND}) measured along L1–L4 is shown in Fig. 8. It can be seen that fluctuations of ρ_{GND} along L2 (MAGBs) and L3 (LAGBs) are more obvious, which indicates that ρ_{GND} on both sides of the interface changes significantly during the staged elimination process of IGBs. Related studies showed that dissociation of HAGBs into LAGBs [43] and annihilation of GBs [44] were both achieved by dislocation emission. The essence was that the full dislocations on the GBs were dissociated into partial dislocations and emitted into the grain, i.e., the new dislocation pairs with opposite Burgers vector were created, then the partial dislocations cancelled out and annihilated with the partial dislocations within the grains or at GBs, i.e., the annihilation of two dislocation pairs was produced by perfect counteracting of the two-opposite-pair Burgers vectors of these two dislocation pairs. Based on the GND distribution results (Fig. 8) at the interface, it can be reasonably inferred that the elimination of IGBs of 6061 Al alloy during HRB



Fig. 8. Misorientation angle and GND density (ρ_{GND}) measured along the lines 'L1' (a), 'L2' (b), 'L3' (c), and 'L4' (d) marked in Fig. 7.

was also achieved by dislocation emission and annihilation. However, due to the effect of CDRX, a large number of dislocations were generated during the thermal deformation of 6xxx Al alloy, resulting in the formation of a large number of substructures. This indicated that the formation and annihilation of GBs (HAGBs) and subgrain boundaries (LAGBs and MAGBs) during HRB was a dynamic process, but the overall trend was that the total amount of GBs and subgrain boundaries was increasing. During this process, the staged elimination of IGBs contributed to interface healing. Pole figures and orientation distribution function (ODF) sections at the hot-rolled interface are shown in Fig. 9. It can be seen that there was a very obvious F texture at the hot-rolled interface.

3.3. Mechanical properties of the joints

The engineering stress-strain curves of the bonding joints with different vacuum is shown in Fig. 10, and ultimate tensile strength (UTS) and total elongation (TE) extracted from Fig. 10 are listed in Table 3. It can be seen that UTS and TE of the hot-rolled and aged joints increased with the packaging vacuum. Tensile fracture morphology of the bonding joints (the tensile frontal and profile fracture characteristics are located in the upper right and lower left corners of the images, respectively) at 10⁵ Pa and 1 Pa is shown in Fig. 11. At 10⁵ Pa, the macroscopic fractures of both hot-rolled and aged joints are relatively flat, with locally smooth areas and small and shallow dimples on the fracture surfaces (Fig. 11(a) and (b)), which indicates weaker bonding effects are obtained. And the EDS results indicate that there is significant oxidation on the fracture surface of the hot-rolled joint at 10⁵ Pa. At 1 Pa, both hot-rolled and aged joints underwent significant deformation during the tensile process, with the fracture sur-



Fig. 9. Pole figures (a) and ODF sections ($\varphi_2 = 0^\circ, 45^\circ, 65^\circ$) (b) at the hot-rolled interface at 10^{-2} Pa.

face showing large and deep dimples (Fig. 11(c) and (d)). In addition, there are a large number of second phase particles in the dimples of hot-rolled and aged fractures. The EDS results indicate that the particle in the hot-rolled matrix is Mg₂Si phase, and the insoluble impurity phase particle in the aged matrix is Al₆(Mn, Fe) phase. Based on the corresponding tensile curves, it can be seen that under 10⁵ and 10³ Pa conditions, the elongation of the bonded joint is significantly lower, and the fracture mode is brittle fracture. After the vacuum reaches 1 Pa, UTS and TE of the corresponding hotrolled joints are 163 MPa and 44.9%, respectively. After solution and aging treatment, the strength of the joint increases to 306 MPa, and the corresponding plasticity decreases to 20.9%. The interface still showed good metallurgical bonding effect and ductile fracture occurred. The UTS of T6-AA6061 Al alloy obtained by Nagasai et al. [45], Liu et al. [1], and Thakur et al. [46] was 318 MPa, 301 MPa, and



Fig. 10. Stress-strain curves of the hot-rolled (a) and aged (b) joints with different vacuum.

Table 3. UTS and TE extracted from Fig. 10

Variante da ana a / Da	UTS / MPa		TE / %		
vacuum degree / Pa	The hot-rolled	The aged	The hot-rolled	The aged	
10 ⁵	155 ± 16	285 ± 36	17.5 ± 11.3	11.0 ± 6.8	
10 ³	161 ± 12	297 ± 18	34.1 ± 9.7	16.9 ± 5.1	
1	163 ± 10	306 ± 15	44.9 ± 6.5	20.9 ± 3.7	
10^{-2}	166 ± 10	305 ± 13	47.1 ± 4.3	22.5 ± 2.3	

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Fig. 11. Tensile fracture morphology and corresponding EDS analysis of the hot-rolled (a, c) and aged (b, d) bonding joints at 10⁵ Pa (a, b) and 1 Pa (c, d); (e) spectrum 1, (f) spectrum 2, and (g) spectrum 3.

282 MPa, respectively. Therefore, it is necessary to increase the vacuum condition to more than 1 Pa during HRB of 6061 aluminum alloy.

3.4. Interfacial evolution mechanism

Based on the results described above, the interfacial microstructure evolution of 6061 Al alloy during HRB and heattreatment can be schematically summarized, as shown in Fig. 12. At 10^5 Pa, there were significant cracks at the bonding interface (Fig. 12(a)). When the vacuum increased to 10^3 Pa, Mg₂Si particles were precipitated along the locally healed interface. However, due to excessive oxidation, the bonding effect at the interface was still poor (Fig. 12(b)), leading to brittle fracture. As the vacuum exceeded 1 Pa, good metallurgical bonding was achieved at the joints due to the effective reduction of oxidation, the hot-rolled bonding interface formed the non-equilibrium GBs, namely IGBs, and Mg₂Si particles were precipitated along the IGBs due to the decrease in solubility during the subsequent cooling process (Fig. 12(c)). In addition, during the hot deformation, the solute atoms have completely dissolved back into the matrix due to the high temperature (530°C), the bonding interface mainly underwent the evolution of grain structure, i.e., part of IGBs transformed from HAGBs to MAGBs to LAGBs, and eventually disappeared completely (Fig. 12(e)), which contributed to the interface healing. This process was achieved by the dissociation, emission, and subsequent annihilation of dislocations on the IGBs. The trace amount of Al_2O_3 film at the interface eventually evolved into MgO during solution and aging treatment (Fig. 12(d)).



Fig. 12. Schematic of interfacial evolution mechanism: (a) the hot-rolled at 10^5 Pa, (b) the hot-rolled at 10^3 Pa, (c) the hot-rolled over 1 Pa, (d) the aged over 1 Pa, and (e) evolution of grain structure at the interface during HRB over 1 Pa.

4. Conclusions

In this work, mechanical properties and interfacial microstructure evolutions of 6061 Al alloy bonding joints produced by hot-roll bonding based on friction stir welding were investigated. The results can be summarized as follows:

(1) Mechanical properties of the joints produced by HRB increased with the packaging vacuum. At 10^5 and 10^3 Pa, there were obvious inclusion defects at the interface due to excessive oxidation. And the corresponding tensile specimens both fractured at the joints, showing obvious brittle fracture.

(2) After exceeding 1 Pa, metallurgical bonding was achieved at the interface due to a significant reduction in oxidation. And the corresponding hot-rolled and aged specimens both fractured at the matrix with UTS of about 163 and 306 MPa.

(3) After exceeding 1 Pa, local IGBs transformed from HAGBs to MAGBs to LAGBs, and eventually disappeared during HRB, which contributed to the interface healing. This process was achieved by the dissociation, emission, and subsequent annihilation of dislocations on the IGBs.

(4) After exceeding 1 Pa, the hot-rolled interface formed the non-equilibrium GBs. Mg_2Si particles were linearly precipitated along the IGBs during the cooling after rolling, and dissolved into the matrix during solution treatment. The trace amount of Al_2O_3 film remaining at the interface eventually evolved into MgO.

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Conflict of Interest

All authors declare that they have no financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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