Characterization of Low Pressure RF Plasma Heating

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Abstract: Compacts of α -Al₂O₃ and Mo powder were heated in radio-frequency (RF) induced low pressure N₂, H₂, Ar, and their mixture plasma. An optical pyrometer, a radiation pyrometer, and a system called Accufiber Model 310 were used to measure the temperature of compacts heated in the plasma. The results indicate that there are different heat transfer mechanisms from plasma to specimens of different physical properties. The Ar plasma showed the highest heating ability among N₂, H₂, and Ar plasma for Al₂O₃ specimens, whereas the H₂ plasma could heat Mo specimens to a higher temperature than the Ar plasma did, even under the same generating conditions.

Key words: plasma heating; temperature measurement; heat transfer mechanism

The heat transfer process from plasma to specimens is much more complicated than in the case that specimens are heated in an ordinary high temperature atmosphere. Some experimental investigation and theoretical analysis of plasma-wall heat transfer have been reported^[1-4]. The existence of free electrons, ions, and neutral species in a plasma and the steep gradients of the plasma parameters, particularly in the vicinity of walls, give rise to a number of peculiar effects which are still poorly understood. Generally, examination of a heat transfer process is based on the measurement of heating procedure. However, the accurate measurement of specimen temperature heated in plasma is very difficult. For examples, some significant errors could be introduced into an optical pyrometric measurement, because of the interference from the luminous plasma and the uncertain emissivity of the specimen surface; a RF plasma operation causes a strong electric noise which will interfere the measurement using a bimetallic thermocouple element.

In this study, a variety of methods for temperature measurement was used, such as by an optical and a radiation pyrometers, an Accufiber system, to measure the temperature of Al₂O₃ and Mo compacts of different physical properties, and to give a comprehensive examination on the heating characteristics of low pressure high intensity RF plasma.

1 Experiment

A schematic drawing of the torch part of plasma heat-

ing apparatus is shown in Fig. 1. High-purity N₂, H₂, Ar, and their mixtures were supplied through a gas controlling system into the plasma torch which consists of a water-cooled quartz tube and an induction coil. A 5 MHz power supply capable of 15 kW plate power was used to generate the plasma. Samples can be set at different positions along the axis of the tube by a boron nitride holder. The pressure of the plasma working gas was fixed at 80 Pa.

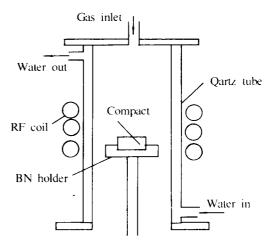


Fig. 1 Schematic view of the plasma heating apparatus

Al₂O₃ and Mo powder compacts of 3 µm grain size, with green densities of about 50% and 68% of their theoretical value respectively, were heated in the plasma and their temperatures were measured by three kinds of methods. The first was to use an optical pyrometer of disappearing filament to measure the equilibrium temperature of the specimens. Errors

caused by the luminous plasma were evaluated, by comparing the measured equilibrium temperature in plasma with the value obtained by extrapolating the cooling curve measured after quenching the plasma. The second was to measure the time variation of specimen temperature, using a radiation pyrometer which can respond to the quick change of the specimen temperature. The third was to use a system called Accufiber Model 310. Fig. 2 shows the schematic drawing of its probe part. This system has a sapphire fiber coated with thin film of platinum alloy and sheathed with a sapphire tube, forming a probe about 3 mm in diameter. The probe can be set into the torch and heated directly by the plasma. Signals from the coated

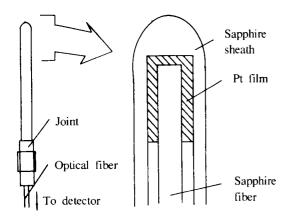


Fig. 2 Schematic view of the probe part of the temperature measurement system called Accufiber Model 310 film are conducted by the fiber to a detector where they are transformed into value of temperature.

2 Results and Discussions

2.1 Heating of Al,O, specimens

Table 1 shows temperatures of Al₂O₃ compacts heated in plasmas for 120s measured with the optical pyrome-

Table 1 Temperature of Al₂O₃ compacts * heated for 120s and equilibrium temperature of sapphire probe**

M-d1	Plasma	Plate	T/°C
Method	gas	power/kW	Temperature / ℃
	N_2	6	1 650
compacts	H_2	6	1 550
	N_2	4.5,6,8	1 200,1 390,1 560
sapphire probe	H_2	6	1 480

Note: * by optical pyrometer and the data were compensated with $\varepsilon = 0.3$;

** by Accufiber Model 310, with the top of the probe set at the same position as the top surface of the compacts

ter, and the equilibrium temperatures of the Accufiber probe. Two things can be noticed from this table. At first, temperatures of the probe rise with the increase of the plate power. Secondly, the probe temperatures reveal that the H₂ plasma exhibited a higher heating ability than the N₂ plasma generated at the same plate power of 6 kW. This could be due to the fact that H₂ has a lower dissociation energy and a higher thermal conductivity than that in the N₂ case. However, the compact heated in the N₂ plasma exhibited a temperature of 1650°C which is 100°C higher than the value of 1550°C reached in the H₂ plasma, which shows inconsistency with the probe measuring.

Fig. 3 shows the time—temperature dependence of the Al_2O_3 compacts measured with the radiation pyrometer. Temperatures passed through a maximum either heated in the N_2 or in the H_2 plasma. But the curves show different "cool-down" characteristics. This indicates that compacts heated in the N_2 and H_2 plasma will exhibit different relative temperature values with the change of time. The maximum temperature of the compact heated in the H_2 plasma is higher than that heated in the N_2 plasma. However, at 120s corresponding to the measured values show in Table 1, the compact heated in the N_2 plasma revealed a temperature higher than that heated in the H_3 plasma. This just

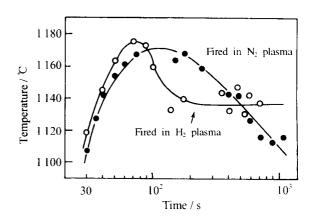


Fig. 3 Time—temperature dependence of Al₂O₃ compacts measured with the radiation pyrometer

clarified the inconsistency of the results shown in Table 1.

E L Kemer, et al. [5] reported that the cool-down phenomenon occurred only in the case of Ar plasma heating but not in N₂, H₂ or other diatomic gas plasma heating. It was found in the present study that not only Ar but also N₂ or H₂ plasma heating of Al₂O₃ compacts exhibited the cool-down effect, with some difference in character when the plasma gas or the specimen

conditions was changed. The time—temperature dependence in Fig. 3 corresponds to different densification behavior shown in Fig. 4. That is, the turning points of the curves in Fig. 3 correspond to the ones of densification rates in Fig. 4, at about 75 s in the case of H₂ plasma heating and at 110 s in the N₂ case of N₂. This behavior could be attributed to the change of heat transfer rate from plasma to compact resulting from the decreasing porosity in compact during sintering^[3].

Among H., N., and Ar plasma, the Ar plasma showed

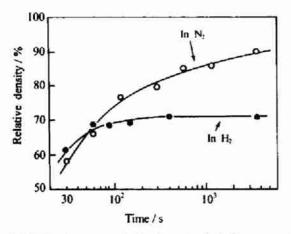


Fig. 4 Relation between relative density of Al₂O₃ compacts and firing time in plasma of 6kW

the highest heating ability for Al₂O₃ compacts. It could even melt the compact surface and the sapphire probe of the Accufiber at a plate power of 8 kW, whereas the H₂ or N₂ plasma could not heat the specimens to that extent at the same plate power. The specimen densified most rapidly obtaining a density of 82% of the theoretical value in the Ar plasma at a plate power of 4.5 kW, compared with those fired in the N₂ and H₃ plasma at a plate power of 6 kW, see Fig. 5. Meanwhile, the specimen surface in Fig. 5(C) fired in the Ar plasma seems to reveal partial melted trace and shows relatively smooth morphology than those fired in the N₂ and H₃ plasma. The different surface morphology in Fig. 5 shows the different interaction effects of the compacts with the three kinds of plasmas.

2.2 Heating of Mo specimens

Table 2 shows the measured temperatures of Mo specimens. The H₂ plasma revealed higher heating ability than Ar plasma for Mo specimens, which is contrary to the heating of Al₂O₃ specimens in the plasma of these two gases. Furthermore, we can notice from Table 2 that the temperature of the Mo compact heated in Ar + 50% H₂ plasma was higher than that heated

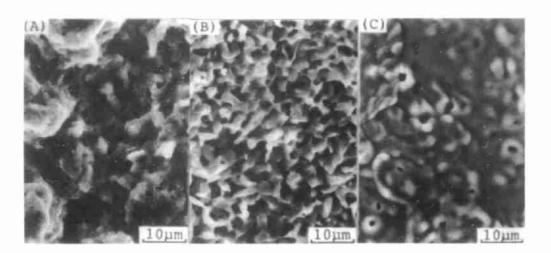


Fig.5 Surface morphologies of Al₂O₃ compacts fired (A) in N₂ plasma of 6kW for 120s obtaining a relative density of 76%, (B) in H₂ of 6kW for 400s obtaining 71%, and (C) in Ar of 4.5kW for 120s obtaining 82%

Table 2 Temperature of Mo compacts measured with the optical pyrometer

Gas type	Temperature / C		
	at 8 kW	at 10 kW	
H ₂	1 730	1 900	
Ar+50% H ₂	1 820	1 900	
Ar	1 610	1 700	

Nate: * as measured without compensation

in any plasma of its component gas at 8 kW.

Fig. 6 shows the density of Mo compacts fired in Ar, H₂, and their mixtures plasma at 10 kW for 180 s. A high heating ability of plasma generally causes rapid densification at the early stage of sintering, although it does not always result in good effect in the whole sintering procedure. Thus, the results in Fig. 6 appear to be consistent with those in Table 2. This could be attributed to that H, has a very low dissociation energy

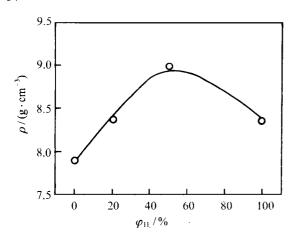


Fig. 6 Relation between sintered density of Mo compact and H, content of Ar+H, plasma at 10kW fired for 180s

(4.48 eV) and its thermal conductivity is much higher than Ar gas in the range up to $2 \times 10^4 \, \text{K}^{[6]}$. The composite effect of the high gas temperature of $\text{Ar}^{[7]}$ and the high thermal conductivity of H_2 caused the high heating ability of the Ar and H, mixture plasma.

2.3 Problems of temperature measurements

The results on the temperature measurement of Al_2O_3 compacts showed that the luminosity of plasma caused an error of $30\sim50^{\circ}$ C higher than the actual temperature of the specimens, when the optical pyrometer was used. The morphology change of specimen surface, according to the different plasma-wall interaction and densification conditions, causes difficulty to estimate the real emissivity of the specimen, which also introduces errors to temperature measurement by the optical pyrometer.

The data in Fig. 3 show that the temperature measured with the radiation pyrometer appears to be about 400°C lower than that measured with the optical pyrometer (see Table 1). The temperature measurements with the optical and radiation pyrometers act in the wavelength range of $650 \sim 1100$ nm. The quartz tube and the cooling water of the plasma torch do not show a significant absorption in this range of wavelength. However, the working mechanism of these two pyrometers is different. In the case that the optical pyrometer is used, it only needs to compare the color of the specimen with that of the filament. And the radiation pyrometer measures the emission energy from a certain area of the specimen. The refraction at the interface of the quartz tube and cooling water might affect the temperature measurement and cause errors, when the radiation pyrometer is used.

The results measured by the Accufiber Model 310 correctly reflected the effects of gas species on the heating characteristics of plasma for Al₂O₃ compacts, but they cannot explain why the plasma heating of Mo compacts is different from that of Al₂O₃ compacts. This could be interpreted to be that the probe of the Accufiber system is sheathed by sapphire which is the same material as Al₂O₃.

2.4 Discussions on heating mechanism

To interpret the different heating characteristics for Mo and Al₂O₃ specimens, some discussions will be given based on the present results and theoretical analysis in other reports.

E Pfender, et al. [3] analyzed the heat transfer of plasma sintering process taking the case of Al₂O₂ as an example. They revealed that heat transfer from plasma to the wall depends strongly on catalytic properties of the surface which results in different boundary layer separating free stream of plasma from the wall. In the case of non-catalytic surface, negative charge on the surface gives rise to an electric field which increases the rate and energy of the ions arriving at the surface, resulting in a substantial increase of the heat transfer. That is, the ion density in plasma greatly affects the heat transfer rate from plasma to specimen of electric insulator. The density of ions in an Ar plasma is generally higher than that in a H₂ or N₃ plasma at the same generating conditions, which result in high heat transfer from Ar plasma to Al,O, compacts. This just interprets the results on the heating of Al,O, specimens in the present work.

On the other hand, G R Chludzinski^[2] determined the heat transfer rate of Ar and Ar + N_2 plasma, by measuring the dT/dt with a inconel-sheathed chromelalumel thermocouple and analyzing the measured results. His results showed that reactions of the atomic recombination are more important than the ion-electron reactions. This suggests that recombination reactions of dissociated atoms in diatomic gas plasma play an important role in enhancing the heat transfer to a specimen of good conductivity, such as inconel in Chludzinski's work and Mo in this study.

Accordingly, E Pfender, et al. and G R Chludzinski analyzed the plasma heating processes based on the temperature measurements of samples of different physical properties. A comprehensive review of the available information on experiment and analysis indicates that we should consider that there are different

heat tran-s-fer mechanisms from plasma to specimen, when the physical properties of heated specimens are different even under the same plasma conditions.

3 Conclusions

The heating characteristics of the low pressure RF plasma for specimens of different physical properties are quite different. The Ar plasma heated Al₂O₃ specimens most effectively among the N₂, H₂, and Ar plasma, whereas the H₂ plasma was more capable to raise the temperature of Mo specimens than the Ar plasma. There could be two mechanisms which dominate the heat transfer rate from plasma to specimen: ion density in plasma greatly affects the heating characteristics when the specimen is an electrical insulator; thermal conductivity, especially the portion caused by the

recombination of dissociated atoms of plasma gas plays an important role in heating the specimens of good electrical conductivity.

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